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Crystal Structure Communications

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5-(2-Hydroxy-4,4-dimethyl-6-oxocyclohex-1-enyl)-3-methyl-2-(methylsulfanyl)-6-phenyl-7*H*-pyrrolo[2,3-*d*]pyrimidin-4(3*H*)-one monohydrate: complex sheets generated by multiple hydrogen bonds

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In the title compound, C₂₂H₂₃N₃O₃S·H₂O, the non-aromatic carbocyclic ring adopts a half-chair conformation. The molecules are linked into complex sheets by a combination of one N-H···O hydrogen bond and three O-H···O hydrogen bonds.

Comment

We have recently reported the preparation of new fused heterocyclic pyrimidine derivatives, such as pyrimido[4,5-b]quinolines, by multicomponent reactions between 6-aminopyrimidine derivatives, 5,5-dimethylcyclohexane-1,3-dione (dimedone) and aryl aldehydes (Quiroga et al., 2006). The extension of this method, with replacement of the aldehyde component by a glyoxal derivative (see scheme), has now provided the title pyrrolo[2,3-d]pyrimidine compound, (I) (Fig. 1), whose molecular and supramolecular structures are reported here.

The bond distances (Table 1) show evidence for strong bond fixation, both within the heterocyclic rings and in the nonaromatic carbocyclic ring; for the atom sequence C51-C56 within this ring, the ring-puckering parameters (Cremer & Pople, 1975) are $\theta = 52.2 (3)^{\circ}$ and $\varphi = 154.4 (4)^{\circ}$. These parameters are very close to the ideal values for the half-chair conformation, viz. $\theta = 50.8^{\circ}$ and $\varphi = (60n + 30)^{\circ}$. Atoms C5, C51, C52, C55 and C56 are almost coplanar, but atoms C53 and C54 deviate from this plane by 0.345 (2) and 0.352 (2) Å, respectively, on opposite sides of the reference plane. The aryl

ring makes a dihedral angle of 16.0 (2)° with the pyrrole ring, while methyl atom C21 is almost coplanar with the adjacent pyrimidine ring.

Within the selected asymmetric unit (Fig. 1), the molecular components are linked by an O-H···O hydrogen bond. These two-component aggregates are linked into complex sheets by a combination of two further O-H···O hydrogen bonds and one N-H···O hydrogen bond (Table 2), each of which, considered in isolation, links pairs of aggregates into centrosymmetric motifs. Each pairwise combination of two

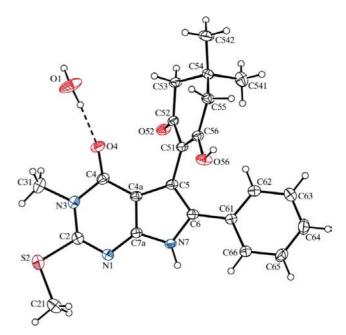


Figure 1 The independent molecular components of (I), showing the atomlabelling scheme and the O-H···O hydrogen bond within the selected asymmetric unit. Displacement ellipsoids are drawn at the 30% probability level.

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such motifs generates a chain of edge-fused rings, and the combination of all three chains generates a complex sheet.

We analyse, firstly, the formation of the three finite zero-dimensional substructures, and then their combinations to form three one-dimensional substructures. Water atom O1 at (x, y, z) acts as a hydrogen-bond donor, via H1B, to carbonyl atom O52 at (1-x, 1-y, 1-z), so generating by inversion an $R_4^4(20)$ (Bernstein et al., 1995) ring centred at $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$, which we denote motif A. Hydroxy atom O56 at (x, y, z) acts as a hydrogen-bond donor to water atom O1 at (-x, 1-y, 1-z), so generating by inversion a second and distinct $R_4^4(20)$ motif, this time centred at $(0, \frac{1}{2}, \frac{1}{2})$, which we denote motif B. Finally, pyrrole atom N7 at (x, y, z) acts as a hydrogen-bond donor to carbonyl atom O52 at (-x, 1-y, -z), so generating by inversion an $R_2^2(14)$ motif centred at $(0, \frac{1}{2}, 0)$, denoted motif C.

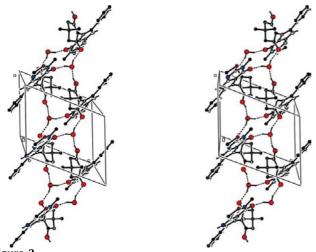
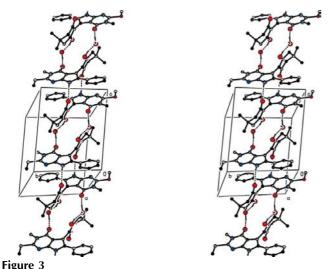
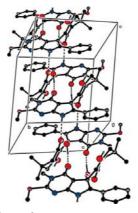


Figure 2 A stereoview of part of the crystal structure of (I), showing the formation of a chain of $R_4^4(20)$ rings along [100] and built from $O-H\cdots O$ hydrogen bonds only. For the sake of clarity, H atoms bonded to C atoms have been omitted



A stereoview of part of the crystal structure of (I), showing the formation of a chain of alternating $R_2^2(14)$ and $R_4^4(20)$ rings along [001] and built from $O-H\cdots O$ and $N-H\cdots O$ hydrogen bonds. For the sake of clarity, H atoms bonded to C atoms have been omitted.



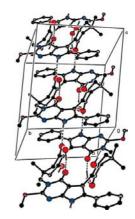


Figure 4
A stereoview of part of the crystal structure of (I), showing the formation of a chain of alternating $R_2^2(14)$ and $R_4^4(20)$ rings along [101] and built from $O-H\cdots O$ and $N-H\cdots O$ hydrogen bonds. For the sake of clarity, H atoms bonded to C atoms have been omitted.

The combination of motifs A and B generates a chain of edge-fused rings, containing two types of $R_4^4(20)$ ring, running parallel to the [100] direction (Fig. 2). The combination of motifs B and C generates a chain of alternating $R_2^2(14)$ and $R_4^4(20)$ rings running parallel to the [001] direction (Fig. 3). Finally, the combination of motifs A and C generates a second chain of $R_2^2(14)$ and $R_4^4(20)$ rings, this time running parallel to the [101] direction (Fig. 4). The combination of any two of the [100], [101] and [001] chains suffices to generate a sheet parallel to (010). There are no direction-specific interactions between adjacent sheets.

Experimental

Equimolar quantities (1 mmol of each component) of 6-amino-3-methyl-2-(methylsulfanyl)pyrimidin-4(3H)-one, 5,5-dimethylcyclohexane-1,3-dione and phenylglyoxal hydrate were mixed, and the mixture was then placed in an open Pyrex-glass vessel and irradiated in a domestic microwave oven for 5 min at 600 W. The product mixture was extracted with ethanol and, after removal of the solvent, the product was recrystallized from ethanol to give crystals of (I) suitable for single-crystal X-ray diffraction (m.p. 565–567 K, yield 45%). MS (EI 70 eV) m/z (%): 410 (27), 409 (M⁺, 100), 395 (19), 394 (75), 311 (27), 284 (43), 264 (13), 236 (9), 88 (19).

Crystal data

 $V = 1056.98 (3) \text{ Å}^3$ $C_{22}H_{23}N_{3}O_{3}S{\cdot}H_{2}O$ $M_r = 427.51$ Z = 2Triclinic, $P\overline{1}$ $D_x = 1.343 \text{ Mg m}^{-3}$ a = 9.11880 (12) ÅMo $K\alpha$ radiation b = 11.3095 (2) Å $\mu = 0.19 \text{ mm}^{-1}$ c = 11.6526 (2) Å T = 298 (2) K $\alpha = 97.5471 (10)^{\circ}$ Lath, colourless $\beta = 110.5868 (10)^{\circ}$ $0.22 \times 0.14 \times 0.10 \text{ mm}$ $\gamma = 104.4677 (11)^{\circ}$

Data collection

Bruker–Nonius KappaCCD diffractometer φ and ω scans Absorption correction: multi-scan (SADABS; Sheldrick, 2003) $T_{\min} = 0.949$, $T_{\max} = 0.982$

26102 measured reflections 4836 independent reflections 3874 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.049$ $\theta_{\rm max} = 27.5^{\circ}$

organic compounds

Refinement

 $\begin{array}{lll} \mbox{Refinement on } F^2 & w = 1/[\sigma^2(F_{\rm o}^2) + (0.0637P)^2 \\ R[F^2 > 2\sigma(F^2)] = 0.057 & + 0.7123P] \\ wR(F^2) = 0.144 & where <math>P = (F_{\rm o}^2 + 2F_{\rm c}^2)/3 \\ S = 1.01 & (\Delta/\sigma)_{\rm max} = 0.002 \\ 4836 \ \mbox{reflections} & \Delta\rho_{\rm max} = 0.71 \ \mbox{e Å}^{-3} \\ 272 \ \mbox{parameters} & \Delta\rho_{\rm min} = -0.59 \ \mbox{e Å}^{-3} \end{array}$

Table 1 Selected bond lengths (Å).

| N1-C2 | 1.302 (3) | C2-S2 | 1.762 (2) |
|---------|-----------|---------|-----------|
| C2-N3 | 1.381 (3) | S2-C21 | 1.786 (3) |
| N3-C4 | 1.414 (3) | C51-C52 | 1.446 (2) |
| C4-C4a | 1.422 (3) | C52-C53 | 1.511 (2) |
| C4a-C5 | 1.427 (3) | C53-C54 | 1.534 (3) |
| C5-C6 | 1.382 (2) | C54-C55 | 1.532 (3) |
| C6-N7 | 1.395 (2) | C55-C56 | 1.494 (2) |
| N7-C7a | 1.353 (2) | C56-C51 | 1.362 (2) |
| C7a-N1 | 1.365 (2) | C52-O52 | 1.236 (2) |
| C4a-C7a | 1.386 (2) | C56-O56 | 1.328 (2) |
| | | | |

Table 2 Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$ | $D-\mathrm{H}$ | $H \cdot \cdot \cdot A$ | $D \cdot \cdot \cdot A$ | $D-\mathrm{H}\cdots A$ |
|--------------------------|----------------|-------------------------|-------------------------|------------------------|
| O1-H1A···O4 | 0.98 | 1.76 | 2.688 (3) | 157 |
| $O1-H1B\cdots O52^{i}$ | 0.98 | 1.90 | 2.762 (3) | 145 |
| $N7-H7\cdots O52^{ii}$ | 0.86 | 2.24 | 2.974 (2) | 143 |
| $O56-H56\cdots O1^{iii}$ | 0.82 | 1.78 | 2.560 (3) | 159 |

Symmetry codes: (i) -x + 1, -y + 1, -z + 1; (ii) -x, -y + 1, -z; (iii) -x, -y + 1, -z + 1

Crystals of (I) are triclinic; the space group $P\overline{1}$ was selected and confirmed by the structure analysis. All H atoms were located in difference maps and then treated as riding atoms, with C–H distances of 0.93 (aromatic), 0.96 (CH₃) or 0.97 Å (CH₂), and O–H distances of 0.82 (hydroxy) or 0.98 Å (water), and with $U_{\rm iso}({\rm H}) = kU_{\rm eq}({\rm C,O})$, where k=1.5 for O-bound and methyl H atoms, and 1.2 for all other H atoms.

Data collection: *COLLECT* (Hooft, 1999); cell refinement: *DENZO* (Otwinowski & Minor, 1997) and *COLLECT*; data reduction: *DENZO* and *COLLECT*; program(s) used to solve structure:

SIR2004 (Burla et al., 2005); program(s) used to refine structure: OSCAIL (McArdle, 2003) and SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97 and PRPKAPPA (Ferguson, 1999).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: SK3043). Services for accessing these data are described at the back of the journal.

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